2.1 INTRODUCTION

Substituted 1,2,4-triazines represent an important class of nitrogen-containing heterocycles. The 1,2,4-triazine core is a versatile synthetic platform to access a widerange of condensed heterocyclic ring system *via* intramolecular Diels-Alder reactions with a vast array of dienophiles. Moreover, the triazines ring system is a key component of commercial dyes, herbicides, insecticides and more recently, pharmaceutical composition. While only a few of these heterocycles are found in nature, they have been prepared and treated as potentially active building blocks in agrochemical and medicinal field. In addition, 1,2,4-triazines have been extensively used as electron deficient dienes for the preparation of pyridine derivatives through their reactions with electron rich dienophiles by the application of Diels-Alder cycloaddition reactions with inverse electron demand.

2.1.1 Review of literatures

1,2,4-triazines are useful intermediates in the synthesis of several heterocyclic systems. They are well established as heterodienes in the inverse electron demand Diels-Alder reaction to form functionalized pyridine derivatives and undergo ring interconversions into five or six-membered aza hetero aromatics when reacted with nucleophile reagent.

The synthesis of 1,2,4-triazines and thier application to the synthesis of pyridine compounds *via* an inverse electron demand Diels-Alder reaction are summarized as follows:

Case (Case, 1965) prepared a number of 1,2,4-triazine derivatives **17a-d**, **19a-g**, **21a-f** and **24** from hydrazidines and α , β -diketones. The synthetic route to 1,2,4-triazines **17a-d**, **19a-g**, **21a-f** and **24** is shown in **Schemes** 9-12, respectively.

Scheme 9 The synthetic route to 3-substituted-5,6-diphenyl-1,2,4-triazines 17a-d

$\underset{H}{\overset{NH}{ \longrightarrow}} \underset{H}{\overset{NH_2}{ \longrightarrow}} +$	Py Py .	EtOH RT	$Py \bigvee_{N} \stackrel{N}{\bigvee_{N}} R$
15	18		19
	Compound		R
	a		2-pyridyl
	b		4-methyl-2-pyridyl
	c		4-ethyl-2-pyridyl
	d		2-quinolyl
	e		2-thiazolyl
	f		4-phenyl-2-pyridyl
	g		2-(1-10-phenanthrolyl)

Scheme 10 The synthetic route to 3-substituted-5,6-di(2-pyridyl)-1,2,4-triazines, 19a-g

Scheme 11 The synthetic route to 1,2,4-triazines 21a-f

Scheme 12 The synthesis of bi-1,2,4-triazine 24

Boger and Panek (Boger and Panek, 1981) reported the short synthetic route for the construction of substituted pyridines **27** as shown in **Scheme** 13. Although this studied led to the development of simple pyridine annulation based on the

regiospecific inverse electron demand Diels-Alder reaction of 1,2,4-triazine 26 with enamine 29, it has two limitations. The first one is the requirement for a preformed the enamine and the other one is the unusual stability of the intermediate when using enamines derived from cyclohexanone. In 1982, they circumvented these difficulties for 1,2,4-triazine or 3-substituted-1,2,4-triazines: 4 Å molecular sieves allowed in situ enamine 29 formation and catalysed the elimination step forming pyridine 27 from dihydropyridine 31, but yield was poor. In addition, they have also prepared the Biaryl CD ring of steptonigrin 36 *via* thermal cycloaddition of 1,2,4-triazine 32 with enamines 33 (Scheme 14).

Scheme 13 Inverse electron demand Diels-Alder reaction of 1,2,4-triazine 27

$$\begin{split} E &= CO_2CH_2CH_3\\ a: R &= R_1 = H\\ b: R &= Me; R_1 = H\\ c: R &= Me; R_1 = 2\text{-}OCH_2C_6H_3\text{-}3,4\text{-}(OCH_3)_2 \end{split}$$

$$H_2N$$
 A
 B
 H_2N
 C
 C
 OH
 OH
 OMC

Scheme 14 The synthesis of biaryl CD ring of steptonigrin **35** *via* thermal cycloaddition of 1,2,4-triazine **32** with enamines **33a-c**

Martin (Martin, 1982) has prepared pyridine C ring of streptonigrin **35** by Diels-Alder reaction of 1,2,4-triazine **38** and aromatic alkyne. The synthetic route to pyridine C ring of streptonigrin was indicated in **Scheme** 15.

Scheme 15 Preparation of pyridine C ring of streptonigrin 35

Boger and Panek (Boger and Panek, 1983) have prepared 1,2,4-triazines **48a-c** via thermal cycloaddition of dimethyl-1,2,4,5-tetrazine-3,6-dicarboxylate **45a-c** with thioimidate. The synthetic route to 1,2,4-triazines **48a-c** was indicated in **Scheme** 16.

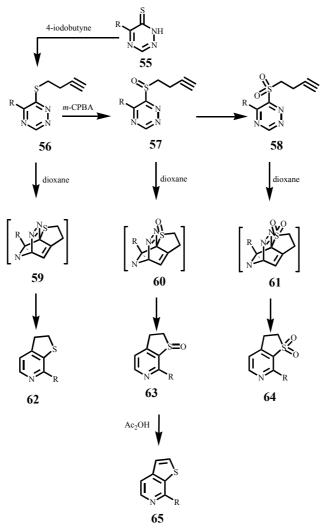
Borwell and Hughes (Borwell and Hughes, 1983) have reported the total synthesis of streptonigrin **35** based on the implementation of two consecutive inverse electron demand Diels-Alder reactions: 1,2,4,5-tetrazine **50** + S-methyl thiomidate **49** (steptronigrin ABC ring construction) and 1,2,4-triazine **51** + morpholino enamine **52** (steptronigrin DE ring construction). The synthetic route to steptronigrin **35** was shown in **Scheme** 17.

Scheme 16 The synthetic route to 1,2,4-triazines **48a-c**

- a) dioxane, tetrazine; b) 4 eq enamine, CHCl₃; c) 12 eq. of PhSeNa, THF-HMBA; MeOH, HCl cat;
- d) 5 eq. of (PhO)₂P(O)N₃, benzene/reflux, 2.5 hrs, H₂O/reflux, 2.5 hrs; e) excess CH₃I, K₂CO₃, THF

Scheme 17 The total synthetic route to steptronigrin 35

Taylor and Macor (Taylor and Macor, 1985) have reported the novel synthesis of thieno[2,3-b]pyridines **62** and thieno[2,3-c]pyridines **65** by using the novel intramolecular Diels-Alder reaction of alkynithio derivative 1,2,4-triazines **56**- **58**. The synthetic route to thieno[2,3-c]pyridines **62**-**65** and thieno[2,3-b]pyridines **62a**- **65a** are shown in **Scheme** 18a and 18b, respectively. Further investigation of this methodology, in 1986, they have also reported the synthesis of highly substituted pyridine compounds **69a-f** and **75** as indicated in **Schemes** 19 and 20.



a: R = Me; b: $R = CH(CH_3)_2$; c: R = Ph

Schemes 18a The synthetic route to thieno[2,3-c] pyridines **62-65**

Schemes 18b The synthetic route to thieno[2,3-b]pyridines 62a-65a

Schemes 19 The synthesis of highly substituted pyridine compounds 69a-f

a) H₂NNHC(SCH₃)=NH₂⁺ Γ ; b) *m*-CPBA; c) Na⁺OCH₂CH₂C CH; d) reflux; e) DDQ

Schemes 20 The synthesis of highly substituted pyridine compounds 75a-e

Benson and co-workers (Benson et al., 1990) have synthesized γ -carbolides **84a-k**, benzo[f][1,7]napthiridines **81a-k** and **82a-k**, or the noncyclized 3-[5-(1,2,4-triazinyl)]indole **78a-k** by the reaction of 1,2,4-triazines **77a-k** and indole **76**. The synthetic pathway was shown in **Scheme** 21.

Ohsumi and Neunhoeffer (Ohsumi and Neunhoeffer, 1992) have prepared the 1,2,4-triazines **88** and **92** with a functional group in the C-6 position. The synthetic route as shown in **Scheme** 22. In addition, in the same year, they have reported the regioselective synthesis of ethyl-1,2,4-triazine-5-carboxylate **100a-i** and the methodology was indicated in **Scheme** 23.

Scheme 21 Synthetic pathway to carbolides **84a-k**, benzo[f][1,7]napthiridines **81a-k** and **82a-k**, or the noncyclized 3-[5-(1,2,4-triazinyl)]ndole **78a-k**

a) TsN₃, Et₃N; b) Ph₃P; c) EtOH, H₂O; d) (R₁CO)₂O; e) AcO⁻⁺NH₄

a) TsN $_3$, Et $_3$ N; b) Ph $_3$ P; c) EtOH, H $_2$ O; d) (R $_2$ CO) $_2$ O; e) AcO $^-$ NH $_4$ $R_1=R_2= alkyl \ or \ phenyl \ groups$

Scheme 22 The synthetic route to 1,2,4-triazines **88** and **92** with a functional group in the C-6

a) CICOCO ₂Et, Et ₃N, Et ₂OH; b) Ph ₃P, Et ₂OH; c) EtOH, H ₂O; d) R ₂C(OMe) ₂NMe ₂; e) $AcO^{-1}NH$ ₄, AcOH;

f) PhC(OEt)₃, AcOH

$$R_1$$
 R_2 X R_1 R_2 R_2 R_3 R_4 R_5 R_5

Scheme 23 The regioselective synthesis of ethyl-1,2,4-triazine-5-carboxylate 100a-i

Pendrak and co-workers (Pendrak et al., 1994) have published the synthesis of mappicine ketone **100** (MPK) analog. The Pyridine-lactam intermediate **104**, which contain the B and C rings of MPK, could be prepared from 1,2,4-triazine **103** *via* an inverse electron demand intramolecular Diels-Alder reaction. The synthetic route to MPK **100** is indicated in **Scheme** 24.

a) H₂S, Et₂NH, PhCH₃; b) H₂NNH₂, 2,3-butanedione, EtOH; c) yeast lipase, allyamine, Hexane/CCl₄; d) xylene; e) diethyl 1,3-acetonedicarboxylate, xylene; f) piperidine, DMF; g) conc. HCl; h) 37% aqueous formaldehyde, thiophenol, AcOH, piperidine, EtOH; i) Raney nikel, EtOH; j) Tf₂NPh, DMF; k) butyl vinyl ether, Et₃N, Pd(OAc)₂, dppd, CH₃CN; l) AcOH, HCl

Scheme 24 The synthetic route to MPK 110

Rykowsky and Wolinsk (Rykowsky and Wolinsk, 1996) have described a novel route to functionalized 3-aminopyridazines **113** by ring opening and ring closure

reaction of 1,2,4-triazines **111a-c** with carbon nucleophiles being a cyano substituent **112a-c** at a carbanionic center. A novel route to functionalized 3-aminopyridazines **113a-c** is shown in **Scheme** 25.

Ph N N X + NC R b)
$$H_3O^+$$

Ph N NH

Ph N NH

R CN

Ph N NH

Ph N NH

R CN

Ph NH NH

Ph N NH

Scheme 25 A novel route to functionalized 3-aminopyridazines 113a-c

Mamolo and co-workers (Mamolo et al., 2000) have synthesized 4*H*-1,2,4-triazine-5-one derivatives **126** and tested for their in *vitro* antimycobaterial activity. Some triazines from their synthesis showed interesting activity against a strain of Mycobacterium tuberculosis. The synthetic route to 4*H*-1,2,4-triazine-5-one derivatives **126** was shown in **Scheme** 26.

Scheme 26 The synthetic route to 4*H*-1,2,4-triazine-5-one derivatives **126a-v**

Stanforth and co-workers (Stanforth et al., 2002) reported the synthesis of pyridine compounds **131a-d** by using aza Diels-Alder reaction. As shown in **Schemes** 27, the triazines **129** and **132** were converted into their corresponding pyridines **131** and **135** in aza Diels-Alder reaction with 2,5-norbornadiene or with 2,3-dihydrofuran, respectively.

Scheme 27 The synthesis of pyridine compounds **131a-d** and **135** by using aza Diels-Alder methodology

Branosawa and co-workers (Branosawa et at., 2002) have synthesized cycloalkenopyridines **139** and **143** *via* the tandem vicarious nucleophilic substitution (VNS) and intramolecular inverse electron demand Diels-Alder reaction on 1,2,4-triazines **137** and **141**. Some of them are compound with significant importance. They processed potent antishock, tuberculostic and antimalarial properties. The synthetic route to cycloalkenopyridines **139** and **143** was indicated in **Scheme** 28.

Ph N SMe

136

CICH₂SO₂R

KOH/DMSO

PhO₂S

Ph N SMe

141

a) NaH/DMF
b) I—(CH₂)nCH₂C
$$\equiv$$
 CH

PhO₂S

Ph N SMe

141

 $n = 1 \text{ or } 2$

PhO₂S

Ph N SMe

142

A

PhO₂S

Ph N SMe

143

PhO₂S

Ph N SMe

144

A

PhO₂S

Ph N SMe

142

A

PhO₂S

Ph N SMe

142

Scheme 28 The synthetic route to cycloalkenopyridines 139 and 143

Further investigation come from Stanforth and co-workers, in 2003 they have reported the preparation of a series of functionalized 2,2'-bipyridines **147a-c** from 1,2,4-triazine **146a-c** precursors by using the aza Diels-Alder methodology (Stanforth et al., 2003). The synthetic route to 2,2'-bipyridines **147a-c** was shown in **Scheme** 29.

Scheme 29 The synthetic route to 2,2'-bipyridines **147a-c**

Lahue and co-workers (Lahue et al., 2003) have published the intramolecular inverse electron demand cycloaddition of 2-substituted amidazoles **148** with various 1,2,4-triazines **149a-b** to produce both imidazo[4,5-c]pyridine (3-deazapurines) **153a-b** and pyrido[3,2-d]pyridine-4-ones(8-deazapurines) **155a-b** was shown in **Scheme** 30.

Charuchin and co-workers (Charushin et al., 2003) have decribed the reaction of 1,2,4-triazines **156a-g** with enamine **157a-g** in which the reactivity of C=C double bond was reduced by electron withdrawing substituents. From this reaction, they found that the reaction of 5,6-unsubstituted 3-aryl-1,2,4-triazines with amino vinyl ketones and aminovinyl esters in acetic anhydride proceeds regioselectively and smoothly at

room temperature resulting in the formation of pyrolo[3,2-e]1,2,4-triazines **158a-g**. The synthetic route to pyrolo[3,2-e]1,2,4-triazines **158a-g** was indicated in **Scheme** 31.

Scheme 30 The synthesis of 3-deazapurines 153a-b and 8- deazapurines 155a-b

OEt

p-NO₂-C₆H₄ d SCh₂Ph OEt

c

Ph e Me

f Ph Ph

 SC_2H_5 OEt g

Scheme 31 The synthetic route to pyrolo[3,2-e]1,2,4-triazines 158a-g

Zhao and co-worker (Zhao et al., 2003) have prepared 1,2,4-triazines **161a-h** by the application of microwave technology as indicated in **Scheme** 32. In the same year, they have also reported a one pot microwave-mediated synthesis of the basic cathine skeleton 163 (Lindsley et al., 2003). The synthetic route to access the cathine skeleton 163 utilized indole as a dienophile and 1,2,4-triazine 162 as a diene in an intramolecular inverse electron demand Diels-Alder reaction. The synthetic route to cathine 163 was indicated in Scheme 33.

R

Scheme 32 The synthesis of 1,2,4-triazines **161a-h** by the application of microwave technology

Scheme 33 One pot microwave-mediated synthesis of basic cathine skeleton 163

Adam and co-workers (Adam et al., 2003) have reported several new approach to the amino 1,2,4-triazine, **169** (GW356194) and identified an efficient approach, based on the use of thiosemicarbazide **167** as a key building block for the synthesis of core heterocycle. The synthetic approaches to **169** (GW356194) were shown in **Schemes** 34 and 35.

CI MeS NH₂ CI CI CI CI CI CI CI NH₂ A MeS NH₂ NH₂ 166
$$C$$
 CI CI CI CI CI NH₂ 166 C CI CI CI CI CI NH₂ 166 C CI NH₂ NH₂

a) EtOH/reflux; b) (POCl $_2$ O, 1,4-dioxane; c) propan-1-ol; d) EtOH/reflux; e) i. oxalyl chloride, 1,4-dioxane, ii. NH $_3$, propan-2-ol; f) i. m-CPBA, DMF, EtOAc, ii. NaBH $_4$, EtOH

Schemes 34 The synthetic approaches to 169 (GW356194) (route A)

a) NaOH/reflux; b) i. NaOH, EtOH, H₂O₂, ii. conc.HCl; c) PCl₅, toluene; d) NH₃, THF, toluene; e) MeI, NaOH, EtOH

Schemes 35 The synthetic approaches to GW356194 169 (route B)

Ibrahim and co-workers (Ibrahim et al., 2003) reported the synthetic approaches towards new condensed thienopyridine ring systems including furo[2,3-b]thieno[3,2-e]pyridines, bisthieno[2,3-b:3',2'-e]pyridines, 5H-chromeno[2,3-b]thieno[3,2-e]pyridines, 5H-benzo(f)chromeno[2,3-b]thieno[3,2-e]pyridines by using the application of intramolecular [4+2] cycloaddition reactions of suitably designed thieno[2,3-e][1,2,4]triazines 181a-e, 183a-e and 185a-e tethered with alkene or alkyne terminals. The synthetic route to triazine derivatives 182a-e, 184a-e and 186a-e was shown in Scheme 36.

a) NaOEt
b)
$$(CO_2EI)_2$$
Ar
$$176$$
b) $(CO_2EI)_2$
Ar
$$178$$
Ar
$$178$$
Ar
$$178$$
Ar
$$178$$
Ar
$$181$$
Ar
$$181$$
Ar
$$181$$
Ar
$$181$$
Ar
$$181$$
Ar
$$182$$
Ar
$$184$$
Ar
$$185$$
Ar
$$186$$
Ar

Scheme 36 The synthetic route to 1,2,4-triazine derivatives 182a-e, 184a-e and 186a-e

Limanto and co-workers (Limanto et al. 2003) described a successful regioselective synthetic approach to 5-substituted-3-amino-1,2,4-triazine 189. The

reaction involves a nucleophilic displacement of readily available α , α -dihalocarbonyl compounds **187a-h**, followed by a condensation of the resulting crude ketonaminals **188a-h** with aminoguanidine in the presence of AcOH in MeOH. The regionselective synthetic route to 5-substituted-3-amino-1,2,4-triazine **189a-h** is indicated in **Scheme** 37.

Scheme 37 The regioselective synthetic route to 5-substituted-3-amino-1,2,4-triazine

189a-h

Alphonse and co-workers (Alphonse et al., 2004) reported a general approach to selective functionalization of 1,2,4-triazines **191** and **193** by using the combining

addition reaction and palladium catalyzed cross-coupling reaction of organometallics with 3-methyl sulfenyl-1,2,4-triazines **190** and **192** as shown in **Scheme** 38.

2.2 eq. R-B(OH)₂, 2.2 eq CuMeSal

$$\begin{array}{c}
N \\
N \\
N
\end{array}$$
190

191

$$\begin{array}{c}
191 \\
N \\
N \\
R \\
R_1 \\
R_1 \\
R_1 \\
R_1 \\
R_2 \\
R_1 \\
R_2 \\
R_3 \\
R_4 \\
R_1 \\
R_2 \\
R_3 \\
R_4 \\
R_1 \\
R_2 \\
R_3 \\
R_4 \\
R_4 \\
R_5 \\
R_5 \\
R_7 \\
R_7 \\
R_8 \\
R_9 \\
R_9$$

2.2 eq. R-SnBu₃, 2.2 eq CuBrMeS

SMe Pd(PPh₃)₄ (5-10 % mol), THF or DME

192

$$X = 0$$
, S

 $X = 0$, S

Scheme 38 Palladium catalyzed cross-coupling reaction of organometallics with 3-methyl sulfenyl-1,2,4-triazines **191** and **193**

Raw and Taylor (Raw and Taylor 2004) have published the synthesis of highly substituted pyridines **195a-k** *via* tethered imine-enamine (TIE) methodology.

This methodology converted 1,2,4-triazines **194a-k** into highly substituted pyridines **195a-k** *via* the inverse electron demand Diels-Alder reaction which avoided the need for a discrete aromatization step (**Scheme** 39). The TIE methodology has also been applied in one pot reaction cascades involving 1,2,4-triazines **196** and utilizing MnO₂-mediate tandem oxidation processes (TOPs) (**Scheme** 40).

Scheme 39 The TIE one pot cascades reaction of 1,2,4-triazines 195a-k

Scheme 40 The TOP- TIE approaches to dihydropyridines and pyridines 198

Further investigation of the chemistry of 1,2,4-triazines, they have also reported an operationally simple method for the construction of complex polycyclic system **200a-d** in a cycloaddition cascade sequence from 1,2,4-triazines **199a-d** (**Scheme** 41) and a one pot cascade reaction sequence from 1,2,4-triazines **201** which allowed formation of 4,5-dihydroazocins **204a-e** (**Scheme** 42).

Scheme 41 Cascade cycloaddition reaction of 1,2,4-triazines to complex polycyclic system **200a-d**

Scheme 42 One pot cascade reaction sequence from 1,2,4-triazines to 4,5-dihydroazocins **204a-e**

The objective of this project is optimizing synthetic methods to 1,2,4-triazines and the use of these compounds in a new route to prepare highly substituted pyridines.

2.2 EXPERIMENTAL

2.2.1 Instruments and Chemicals

¹H and ¹³C NMR spectra were recorded on Jeol EX-270 and EX-400 instruments running at 270 MHz and 400 MHz for proton and 68 MHz and 100 MHz carbon nuclei, respectively using residual solvent peaks as the internal standard. Coupling constants were given in Hertz and ¹³C spectra were verified using DEPT experiments. Microwave reactions were carried out in a CEM Discover Microwave Synthesizer. Melting points were recorded on an Electrothermal IA9100 digital melting point apparatus and were uncorrected. Infrared spectra were recorded on an ATI Mattson Genisis FT-IR spectrophotometer using NaCl plates. Low resolution electron impact (EI) mass spectra were recorded on a Kratos MS 25 spectrometer. Chemical ionization (CI) and high resolution mass spectra were recorded on a Micromass Autospec spectrometer. Flash column chromatography was performed using Matrex silica gel 60 (70-200 µm) and the eluent specified. PE is the fraction of petroleum ether boiling in the range 40-60 °C, EtOAc is ethyl acetate, DCM is dichloromethane and EtOH is ethanol which were stored over 4 Å molecular sieves. Except where specified, all reagents were purchased from commercial sources and were used without further purification.

2.2.2 Synthesis of 2-pyridylamidrazone 205

$$\bigcap_{N} \bigcap_{H} \bigcap_{N \to 1} NH_2$$

A mixture of 2-cyanopyridine (5.346 g, 51 mmol), EtOH (9 mL) and hydrazine monohydrate (15 mL) was stirred at room temperature for 3 hours, after which water (9 mL) was added and the reacting mixture was extracted with ether (3 x 20 mL). Upon extraction a white solid precipitated which was filtered and stored. The filtrate was dried over anhydrous MgSO₄, and the ether was removed under reduced pressure. The residue was combined with the white solid and recrystallized from hot toluene to give 2-pyridylamidrazone **205** as a white solid (3.091 g, 44%), mp 95.5-96.0 °C; R_f 0.20 (MeOH-EtOAc, 1:9); δ_H (400 MHz, CDCl₃) 8.15 (1H, d, J = 4.0 Hz, H-6), 7.93 (1H, d, J = 7.9 Hz, H-3), 7.61 (1H, dd, J = 1.2, 7.9 Hz, H-4), 7.20 (1H, dd, J = 1.2, 7.9 Hz, H-5), δ_C (100 MHz, CDCl₃) 150.1 (q), 148.0 (q), 147.9 (CH), 136.3 (CH), 123.7 (CH), 119.6 (CH).

2.2.3 Synthesis of 1-cyclohexyl-2-hydroxyketone 206b

2-Cyclohexylmethylketone (0.4 mL, 3 mmol) was added to a stirred

solution of trifluoroacetic acid (0.4 mL, 5 mmol), water (5 mL) and CH₃CN (10 mL). Then [bis(trifluoroacetoxy)iodo]benzene (2.22 g, 5 mmol) was added and the solution was heated at reflux for 3 hours. The reaction was monitored by TLC. When the reaction was complete the reaction mixture was cooled and concentrated *in vacuo* to remove the CH₃CN. The residue was partitioned with DCM (50 mL) and water (10 mL) and the aqueous phase was extracted with DCM (3 x 20 mL). The combined organic extracts were then washed with a saturated NaHCO₃ (3 x 20 mL), dried over anhydrous MgSO₄, filtered and concentrated *in vacuo*. The crude product was purified by flash column chromatography (100% DCM) to give the title compound **206b** as a colorless oil (0.299 g, 55%); R_f 0.40 (PE-EtOAc, 1:1); δ_H (400 MHz, CDCl₃) 4.23 (2H, s), 2.31 (1 H, tt, J = 3, 11.3 Hz), 1.78-1.64 (4H, m), 1.63-1.60 (1H, m), 1.49-1.13 (5H, m).

2.2.4 Synthesis of 1-hydroxy-2-heptanone 206c and 3-hydroxyl-2-heptanone 206d

2-Heptanone (0.7 mL, 5 mmol) was added to a stirred solution of trifluoroacetic acid (0.6 mL, 8 mmol), water (5 mL) and CH₃CN (15 mL). [bis(Trifluoroacetoxy)iodo]benzene (3.45 g, 8 mmol) was added and the solution heated at reflux for 3 hours. The reaction was monitored by TLC. When the reaction was complete the reaction mixture was cooled and concentrated *in vacuo* to remove the CH₃CN. The residue was partitioned with DCM (50 mL) and water (10 mL) and the aqueous phase was extracted with DCM (3 x 20 mL). The combined organic extracts were then washed with a saturated NaHCO₃ (3 x 20 mL), dried over

anhydrous MgSO₄, filtered and concentrated *in vacuo*. The crude product was purified by flash column chromatography (100% DCM) to give the mixture of **206c** and **206d** as colorless oils (0.38 g, 58%, ratio 1.6:1 by 1 H NMR); R_f 0.50 (PE-EtOAc, 1:1); $\delta_{\rm H}$ (400 MHz, CDCl₃) **206c**: 4.16 (2H, s), 2.32 (2H, t, J = 7.6 Hz), 1.55 (2H, quin, J = 7.6 Hz), 1.22 (4H, m), 0.81 (3 H, t, J = 7.6 Hz); **206d**: 4.10 (1H, dd, J = 3.7, 7.3 Hz), 2.11 (3H, s), 1.79-1.71 (1H, m), 1.49-1.16 (7H, m), 0.82 (3H, t, J = 7.0 Hz).

2.2.5 Synthesis of 1-(2-furyl)-2-hydroxyketone 206f

2-Furylmethylketone (0.566 g, 5.1 mmol) was added to a stirred solution of trifluoroacetic acid (0.4 mL, 5 mmol), water (5 mL) and CH₃CN (10 mL). [bis(Trifluoroacetoxy)iodo]benzene (2.35 g, 5.4 mmol) was added and the solution was refluxed for 3 hours. The reaction was monitored by TLC. When the reaction was complete the reaction mixture was cooled and concentrated *in vacuo* to remove the CH₃CN. The residue was partitioned with DCM (50 mL) and water (10 mL). The aqueous phase was extracted with DCM (3 x 20 mL). The combined organic extracts were then washed with a saturated NaHCO₃ (3 x 20 mL), dried over MgSO₄, filtered and concentrated *in vacuo*. The crude product was purified by flash column chromatography (100% DCM) to give the title compound **206f** as a white solid (0.162 g, 25%), mp 79.8-80.4 °C; R_f 0.33 (PE-EtOAc, 1:1); δ_H (400 MHz CDCl₃) 7.56 (1H, d_3 J = 1.2 Hz), 7.23 (1H, d_4 J = 1.2, 1.2 Hz), 6.53 (1H, d_3 J = 1.2 Hz), 4.67 (2 H, s).

2.2.6 Synthesis of amidrazone 207a

To a solution of 2-pyridylamidrazone **205** (0.457 g, 3.3 mmol) in EtOH (5 mL) was added hydroxy acetone (0.100 mL, 1.4 mmol) and the mixture stirred at room temperature for 4 hours. The reaction mixture was concentrated *in vacuo* and purified by flash column chromatography (MeOH- EtOAc, 1:9) to give the title compound **207a** as a yellow solid (0.262 g, 93%), mp 94.5-95.6 °C; R_f 0.25 (MeOH- EtOAc, 1:9); V_{max} (film/cm⁻¹) 3500, 3375, 1630, 1587, 1474, 1072 and 801; δ_{H} (400 MHz, CDCl₃) 8.43 (1H, d, J = 4.6 Hz, pyridyl), 8.18 (1H, d, J = 8.0 Hz, pyridyl), 7.61 (1 H, dd, J = 8.0, 8.0 Hz, pyridyl), 7.20 (1H, dd, J = 4.6, 8.0 Hz, pyridyl), 4.17 (2H, s), 1.95 (3H, s, CH₃); δ_{C} (100 MHz, CDCl₃) 163.0 (q), 154.1 (q), 150.4 (q), 148.2 (CH), 136.4 (CH), 124.9 (CH), 121.0 (CH), 65.3 (CH₂), 14.7 (CH₃); m/z (CI):193 ([M+H]⁺, 100%) [HRMS (CI): calcd. for $C_9H_{13}N_4O$, 193.1089. Found: [M+H]⁺, 193.1088.

2.2.7 Synthesis of amidrazone 207b

$$HO$$
 H_2N
 N
 N

2.2.8 Synthesis of amidrazone 207c and 207d

To a solution of 2-pyridylamidrazone **205** (0.400 g, 2.9 mmol) in EtOH (3 mL) was added the mixture of 1-hydroxy-2-heptanone **206c** and 3-hydroxyl-2-heptanone **206d** (0.136 g, 1.0 mmol). The mixture was stirred at room temperature for 2 hours and the reaction mixture concentrated *in vacuo* and purified by flash column chromatography (PE-EtOAc, 1:1) to give the title compounds **207c** and **207d** as a

yellow viscous oil which were inseparable by column chromatography (0.207 g, 80%), R_t 0.5 (PE-EtOAc, 1:1).

2.2.9 Synthesis of amidrazone 207e

$$HO \longrightarrow H_2N \longrightarrow N$$

To a solution of 2-pyridylamidrazone **205** (0.250 g, 1.8 mmol) in EtOH (3 mL) and DCM (3 mL) was added 1,3-dihydroxyketone (dimer) **206e** (0.225 g, 1.2 mmol) and the mixture was stirred at room temperature for 3 hours. The reaction mixture was concentrated *in vacuo* and purified by flash column chromatography (MeOH- EtOAc, 1:9) to give the title compound **207e** as a yellow solid (0.413 g, 80%), mp 117.8-118.7 °C; R_f 0.46 (MeOH- EtOAc, 1:9); V_{max} (film/cm⁻¹) 3374, 1627, 1586, 1564, 1470, 1057 and 742; $\delta_{\rm H}$ (400 MHz, methanol- d_3) 8.59 (1H, d, J = 4.3 Hz, pyridyl), 8.12 (1H, d, J = 8.0 Hz, pyridyl), 7.84 (1H, ddd, J = 1.8, 8.0, 8.0 Hz, pyridyl), 7.44 (1H, dd, J = 4.3, 8.0 Hz, pyridyl), 4.63 (2H, s), 4.44 (2H, s); $\delta_{\rm C}$ (100 MHz, methanol- d_3) 166.3 (q), 157.0 (q), 151.8 (q), 149.6 (CH), 137.9 (CH), 126.4 (CH), 122.3 (CH), 63.2 (CH₂), 60.5 (CH₂); m/z (CI) 209 ([M+H]⁺, 76%), 122 (100) [HRMS (CI): calcd for $C_9H_{13}N_4O_2$, 209.1038. Found: [M+H]⁺, 209.1035.

2.2.10 Synthesis of 3-(2-pyridyl)-6-methyl-1,2,4-triazine 208a

Microwave heating: To a 10 mL CEM Discover reaction vial with a stir bar was placed amidrazone $\mathbf{207a}$ (0.096 g, 0.5 mmol), toluene (0.5 mL) and \mathbf{MnO}_2 (0.228 g, 2.6 mmol). The reaction vessel was irradiated for 5 minutes at 120 °C (power 200 W, pressure up to 300 psi). After 5 minutes, the vessel was cooled down to 50 $^{\circ}$ C. The crude mixture was then filtered through Celite[®], washed well with DCM and concentrated in vacuo giving the pale yellow solid. In the second step, the yellow solid, along with toluene (1 mL) and glacial AcOH (0.02 mL) were placed in a 10 mL microwave vessel and irradiated at 150 °C for 10 minutes (power 300 W, pressure up to 300 psi) followed by cooling to 50 °C. The homogenous solution was concentrated in vacuo to give the title compound 208a as a red solid (0.084g, 98% without purifying), mp 68.2-69.0 °C; R_f 0.20 (MeOH- EtOAc, 1:9); V_{max} (film/cm⁻¹) 3413, 2960, 1641, 1560, 1408, 1280, 1048 and 769; $\delta_{\!\!\! H}$ (400 MHz, CDCl $_{\!\!\! 3}$) 8.81 (1H, d, J=3.7 Hz, pyridyl), 8.63 (1H, s, H-5, triazine), 8.60 (1H, d, J = 7.9 Hz, pyridyl), 7.86 (1H, ddd, J = 1.8, 7.9, 7.9 Hz, pyridyl), 7.40 (1H, ddd, J = 1.8, 3.7, 7.9 Hz, pyridyl),2.75 (3H, s, CH₃); $\delta_{\rm C}$ (100 MHz, CDCl₃) 161.5 (q), 157.6 (q), 152.6 (q), 150.4 (CH), 149.9 (CH), 137.2 (CH), 125.5 (CH), 123.7 (CH), 19.6 (CH₂); m/z (EI) 172 ([M]⁺, 40%), 144 (47), 105 (100), 78 (32), 51 (24), 39 (28) [HRMS (EI): calcd. for C₀H₈N₄, 172.0748. Found: [M]⁺, 172.0752 (1.9 ppm error)].

Conventional heating: To a solution of aldehyde amidrazone 210a (0.070 g, 0.36 mmol) in DCM (2 mL) was added glacial AcOH (0.01 mL). The reaction mixture

was heated at reflux for 48 hours followed by TLC. The reaction mixture was then concentration *in vacuo* to give the title compound **208a** (0.056 g, 89%).

2.2.11 Synthesis of 3-(2-pyridyl)-6-cyclohexyl-1,2,4-triazine 208b

To a 10 mL CME Discover reaction vial with a stir bar was placed amidrazone 207b (0.157 g, 0.6 mmol), toluene (0.8 mL) and MnO₂ (0.192 g, 2.2 mmol). The reaction vessel was irradiated for 15 minutes at 150 °C (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled down to 50 °C and the crude mixture filtered through Celite and the residue washed well with DCM and concentrated in vacuo giving the title compound 208b as a dark yellow solid (0.138 g, 97%, without further purification), mp 108.7-109.0 °C; R_f 0.40 (MeOH-EtOAc, 1:9); V_{max} (film/cm⁻¹) 2927, 2851, 1624, 1564, 1498, 1470, 1443, 1046, 989, 773 and 743; $\delta_{\rm H}$ (400 MHz, CDCl₃) 8.82 (1H, d, J = 4.9 Hz, pyridyl), 8.64 (1H, s, H-5, triazine), 8.63 (1H, d, J = 8.0 Hz, pyridyl), 7.88 (1H, ddd, J = 1.5, 8.0, 8.0 Hz, pyridyl), 7.43 (1H, dd, J = 4.9, 8.0 Hz, pyridyl), 2.99 (1H, t, J = 8.0 Hz, cyclohexyl), 2.03-1.87 (4H, m, cyclohexyl), 1.78-1.61 (3H, m, cyclohexyl), 1.48-1.20 (3H, m, cyclohexyl); $\delta_{\rm C}$ (100 MHz, CDCl₂) 163.7 (q), 161.0 (q), 151.9 (q), 149.6 (CH), 148.1 (CH), 136.5 (CH), 124.7 (CH), 122.9 (CH), 41.7 (CH), 31.4 (2 x CH₂), 25.4 (2 x CH₂), 24.9 (CH₂); m/z (EI) 240 ($[M]^+$, 76%), 105 (100) [HRMS (EI): calcd. for $C_{14}H_{16}N_4$, 240.1374. Found: $[M]^+$, 240.1380 (2.2 ppm error)].

2.2.12 Synthesis of 3-(2-pyridyl)-6-pentyl-1,2,4-triazine 208c and 3-(2-pyridyl)-5-butyl-6-methyl-1,2,4-triazine 208d

To a 10 mL CEM Discover reaction vial with a stir bar was placed the mixture of amidrazone **207c** and **207d** (0.069 g, 0.27 mmol, ratio 1.6:1), toluene (1.0 mL) and MnO₂ (0.117 g, 1.3 mmol). The reaction vessel was irradiated for 30 minutes at 120 °C (power 200 W, pressure up to 300 psi). After 30 minutes, the vessel was cooled down to 50 °C and the crude mixture then filtered through Celite , washed well with DCM and concentrated *in vacuo* to give the dark yellow viscous oil. In the second step, the dark yellow oil, toluene (1 mL) and glacial AcOH (0.2 mL) were placed in a 10 mL microwave vessel and irradiated at 150 °C for 10 minutes (power 300 W, pressure up to 300 psi). The mixture was then cooled down to 50 °C and the homogenous solution concentrated *in vacuo* to give the crude mixture of **208c** and **208d**. Purification by flash column chromatography (MeOH-EtOAc, 5:95) gave the title compounds **208c** (0.021 g, 55%, calculated from the ratio of **207c**: **207d**) and **208d** (0.015 g, 63%, calculated from the ratio of **207c**: **207d** (1.6:1)) as dark brown viscous oils.

Data for compound **208c**, R_f 0.28 (MeOH- EtOAc, 0.5:9.5); \mathbf{V}_{max} (film/cm⁻¹) 2956, 2929, 2860, 1674, 1586, 1507, 1270, 1118, 771 and 735; δ_{H} (400 MHz, CDCl₃) 8.77 (1H, d, J = 4.6 Hz, pyridyl), 8.76 (1H, s, H-5, triazine), 8.57 (1H, d, J = 7.9 Hz, pyridyl), 7.82 (1H, ddd, J = 1.8, 7.9, 7.9 Hz, pyridyl), 7.37 (1H, dd, J = 4.6, 7.9 Hz, pyridyl), 2.97 (2H, t, J = 7.6 Hz, CH₂), 1.76 (2H, quin, J = 7.6 Hz), 1.30-1.29 (4H, m),

0.81 (3H, t, J = 7.0 Hz, CH₃); $\delta_{\rm C}$ (100 MHz, CDCl₃) 161.6 (q), 161.2 (q), 152.6 (q), 150.4 (CH), 149.7 (CH), 137.2 (CH), 125.5 (CH), 123.7 (CH), 33.5 (CH₂), 31.3 (CH₂), 28.9 (CH₂), 22.4 (CH₂), 14.0 (CH₃); m/z (EI) 228 ([M]⁺, 35%), 172 (88), 105 (100) [HRMS (EI): calcd. for C₁₃H₁₆N₄, 228.1374 Found: [M]⁺, 228.1375.

Data for compound **208d**, R_f 0.23 (MeOH-EtOAc, 0.5:9.5); V_{max} (film/cm⁻¹) 2958, 2931, 1669, 1624, 1587, 1523, 1394, 1049 and 997; δ_{H} (400 MHz, CDCl₃) 8.81 (1H, d, J = 4.6 Hz, pyridyl), 8.53 (1H, d, J = 7.9 Hz, pyridyl), 7.83 (1H, ddd, J = 1.2, 7.9, 7.9 Hz, pyridyl), 7.38 (1H, dd, J = 4.6, 7.92 Hz, pyridyl), 2.86 (2H, t, J = 7.6 Hz), 2.73 (3H, s, CH₃), 1.72 (2H, m), 1.42 (2H, sextet, J = 7.64 Hz), 0.91 (3H, t, J = 7.64 Hz); δ_{C} (100 MHz, CDCl₃) 163.4 (q), 162.3 (q), 157.4 (q),154.0 (q), 151.1 (CH), 137.9 (CH), 126.0 (CH), 124.6 (CH), 35.4 (CH₂), 30.4 (CH₂), 23.6 (CH₂), 20.2 (CH₃), 14.7 (CH₃); m/z (EI) 228 ([M]⁺, 16%), 105 (100) [HRMS (EI): calcd. for C₁₃H₁₆N₄, 228.1374 Found: [M]⁺, 228.1376.

2.2.13 Synthesis of 3-(2-pyridyl)-6-methylenehydroxyl-1,2,4-triazine 208e

Microwave heating: Aldehyde amidrazone **210e** (0.030 g, 0.14 mmol) and toluene (1.0 mL) were placed in a 10 mL microwave reaction tube and irradiated at 150 °C for 15 minutes (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled to 50 °C and checked by TLC. which showed the reaction to be complete. The homogenous solution was concentrated *in vacuo* to give the title compound **208e** as a red solid (0.034 g, 95%, without further purification), mp 246 °C (decomposed); R_f 0.27 (MeOH-EtOAc, 2:1); V_{max} (film/cm⁻¹) 3404, 1634, 1587, 1402,

1073, 1026 and 772; $\delta_{\rm H}$ (400 MHz, methanol- d_3) 9.01 (1H, s, H-5, triazine), 8.75 (1H, d, J = 4.8 Hz, pyridyl), 8.57 (1H, d, J = 7.8 Hz, pyridyl), 8.04 (1H, ddd, J = 1.5, 7.8, 7.8 Hz, pyridyl), 7.6 (1H, ddd, J = 1.5, 4.83, 7.9 Hz, pyridyl), 5.00 (2H, s, CH₂); $\delta_{\rm C}$ (100 MHz, methanol- d_3) 162.8 (q), 161.7 (q), 153.2 (q), 150.7 (CH), 150.2 (CH), 138.9 (CH), 127.1 (CH), 124.8 (CH), 62.4 (CH₂); m/z (EI) 188 ([M]⁺, 6%), 105 (100) [HRMS (EI): calcd. for $C_9H_8ON_4$, 188.0698. Found: [M]⁺, 188.0702.

Conventional heating: To a solution of aldehyde amidrazone 210e (0.021 g, 0.1 mmol) in CHCl₃ (2 mL). The reaction mixture was heated to reflux for 24 hours monitored by TLC. which showed the reaction to be complete. The mixture was then concentration *in vacuo* to give the title compound 208e (0.017g, 89%).

2.2.14 Synthesis of 3-(2-pyridyl)-6-(2-furyl)-1,2,4-triazine 208f

To a 10 mL CEM Discover reaction vial was placed 1-(2-furyl)-2-hydroxyketone **206f** (0.035, 0.3 mmol), amidrazone **205** (0.040 g, 0.3 mmol) and EtOH (0.2 mL). The reaction vessel was irradiated for 40 minutes at 90 °C (power = 200 W, pressure up to 300 psi). After 40 minutes, the vessel was cooled to 50 °C and monitored by TLC. which showed the reaction to be complete. The homogenous solution was concentrated *in vacuo* to give dark yellow viscous oil. In the second step, the dark yellow oil, toluene (1 mL) and MnO₂ (0.052 g, 0.6 mmol) were placed in a 10 mL microwave vessel and irradiated at 120 °C for 30 minutes (power 200 W, pressure up to 300 psi). After 30 minutes, the vessel was cooled to 50 °C and monitored by

TLC. which showed the remaining starting material so a second portion of MnO₂ (0.052 g, 0.6 mmol) was added and irradiated at 150 °C for 5 minutes (power = 200 W, pressure up to 300 psi). After cooling down to 50 °C, the reaction mixture was filter through Celite [®], washed with DCM, concentrated *in vacuo* and purified by flash column chromatography (EtOAc) to give the title compound **208f** as a red solid (0.015 g, 22%), mp 157.8-159.2 °C; R_f 0.19 (EtOAc); V_{max} (film/cm⁻¹) 3054, 2986, 2305, 1639, 1422, 1265, 896, 737 and 706; $\delta_{\rm H}$ (400 MHz, CDCl₃) 9.08 (1H, s, H-5, triazine), 8.81 (1H, d, J = 4.0 Hz, pyridyl), 8.64 (1H, d, J = 7.9 Hz, pyridyl), 7.87 (1H, ddd, J = 1.5, 7.9, 7.9 Hz, pyridyl), 7.65 (1H, brs, furyl), 7.44 (1H, d, J = 3.4 Hz, furyl), 7.44-7.40 (1H, m, pyridyl), 6.62-6.61 (1H, m, furyl); $\delta_{\rm C}$ (100 MHz, CDCl₃) 160.9 (q), 152.2 (q), 150.3 (CH), 149.0 (C), 148.2 (q) 145.5 (CH), 145.2 (CH), 137.1 (CH), 125.5 (CH), 123.7 (CH), 112.9 (CH), 112.8 (CH); m/z (CI) 225 ([M+H]⁺, 100%) [HRMS (EI): calcd. for C₁₂H₆N₄O, 225.0776 Found: [M+H]⁺, 225.0775.

2.2.15 Synthesis of 3-(2-pyridyl)-6-cyclohexyl-1,2,4-triazine 208g

To a 10 mL CEM Discover reaction vial with a stir bar was placed hydroxy acetophenone **206g** (0.040, 0.3 mmol), amidrazone **205** (0.040 g, 0.3 mmol) and EtOH (0.2 mL). The reaction vessel was irradiated for 10 minutes at 90 °C (power 100 W, pressure up to 300 psi). After 10 minutes, the vessel was cooled to 50 °C and the homogenous solution was concentrated *in vacuo* to give dark yellow viscous oil. In the

second step, the dark yellow oil, toluene (1 mL) and MnO₂ (0.026 g, 0.3 mmol) were placed in a 10 mL microwave vessel and irradiated at 120 °C for 15 minutes (power 200 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled to 50 °C and monitored by TLC. which showed that the starting material remained so a second portion of $\mathrm{MnO_2}$ (0.026 g, 0.3 mmol) was added and the mixture irradiated at 120 °C for 15 minutes (power 200 W, pressure up to 300 psi). After cooling to 50 °C, the reaction mixture was filter through Celite[®], washed well with DCM, concentrated in vacuo and purified by flash column chromatography (EtOAc) to give the title compound **208g** as a yellow solid (0.034 g, 50%), mp 164.5-165.5 °C; R_f 0.20 (EtOAc); V_{max} (film/cm $^{-1}$) 1583, 1444, 1400, 1115, 1075, 1043, 987, 775, 740, 729 and 688; $\delta_{\rm H}$ (400 MHz, CDCl₃) 9.14 (1H, s, H-5, triazine), 8.84 (1H, d, J=4.0 Hz, pyridyl), 8.68 (1H, d, J = 8.2 Hz, pyridyl), 8.14 (2H, m, pyridyl and aromatic), 7.89 (1H, ddd, J = 1.8, 8.2, 8.2 Hz, pyridyl), 7.55-7.54 (3H, m, aromatic), 7.40 (1H, ddd, J= 1.8, 4.0, 8.2 Hz, pyridyl); $\delta_{\rm C}$ (100 MHz, CDCl₃) 161.5 (q), 155.9 (q), 152.2 (q), 150.3 (CH), 146.9 (CH), 137.1 (CH), 132.8 (q), 131.1 (q), 129.3 (2 x CH), 126.8 (2 x CH), 125.5 (CH), 123.7 (CH); m/z (EI) 234 ([M] $^+$, 20%), 206 (60) 102 (100) [HRMS (EI): calcd. for $C_{14}H_{16}N_4$, 234.0905. Found: $[M]^+$, 234.0901.

2.2.16 Synthesis of 3-(2-pyridyl)-5-methyl-1,2,4-triazine 209a

To a 10 mL CEM Discover reaction vial was placed amidrazone **205** (0.040 g, 0.3 mmol), hydroxy acetone **206a** (0.022 mL, 0.3 mmol), MnO_2 (0.026 g, 0.3

mmol), powder 4Å molecular sieves (0.100 g) and DCM (0.5 mL). The reaction vessel was irradiated for 80 minutes at 55 °C (power 100 W, pressure up to 300 psi). After 80 minutes, the vessel was cooled to 50 °C and the mixture filtered through Celite® and the residue washed well with DCM and concentrated in vacuo to give dark yellow viscous oil. In the second step, the dark yellow viscous oil and toluene (1 mL) were placed in a 10 mL microwave vessel and irradiated at 120 °C for 50 minutes and cooled down to 50 °C. The homogenous solution was concentrated in vacuo and purified by flash column chromatography (MeOH-EtOAc, 5:95) to give title compound **209a** (0.023, 45%) as a red solid, mp 86.5-87.6 °C; R_f 0.19 (MeOH-EtOAc, 5: 95); V_{max} (film/cm⁻¹) 1633, 1353, 1279, 1252, 1150, 987, 767 and 739; δ_{H} $(270 \text{ MHz}, \text{CDCl}_3) 9.05 \text{ (1H, } s, \text{H-6, triazine)}, 8.79 \text{ (1H, } dd, J = 2.2, 4.8 \text{ Hz, pyridyl)},$ 8.55 (1H, d, J = 8.2 Hz, pyridyl), 7.81 (1H, ddd, J = 2.2, 8.2, 8.2 Hz, pyridyl), 7.37 (1H, dd, J = 4.8, 8.2 Hz, pyridyl), 2.61 (3H, s, CH); $\delta_{\rm C}$ (100 MHz, CDCl₃) 161.8 (q), 157.9 (q), 152.3 (q), 150.3 (CH), 149.8 (CH), 137.1 (CH), 125.4 (CH), 123.6 (CH), 19.5 (CH₂); m/z (EI) 172 ([M]⁺, 17%), 144 (45), 105 (100), 78 (37), 51 (28), 39 (33) [HRMS (EI): calcd. for $C_0H_8N_4$, 172.0748. Found: $[M]^+$, 172.0749.

2.2.17 Synthesis of 3-(2-pyridyl)-5-cyclohexyl-1,2,4-triazine 209b

To a 10 mL CEM Discover reaction vial was placed amidrazone **205** (0.050 g, 0.37 mmol), 1-cyclohexyl-2-hydroxyketone **206b** (0.053 g, 0.37 mmol), MnO_2 (0.032 g, 0.37 mmol) and DCM (0.5 mL). The reaction vessel was irradiated for 30

minutes at 55 °C (power 100 W, pressure up to 300 psi). After 30 minutes, the vessel was cooled to 50 °C and second portion of MnO₂ (0.037 g, 0.38 mmol) and powder 4Å molecular sieves (0.100 g) added. The reaction vessel was irradiated for 50 minutes at 55 °C (power 100 W, pressure up to 300 psi). After 55 minutes, the vessel was cooled to 50 °C, the residue was then filtered through Celite ®, washed with DCM, concentrated in vacuo and purified by flash column chromatography (MeOH-EtOAc, 5:95) to give the title compound **209b** as a yellow solid (0.033 g, 38%), mp 117.4-118.5 °C; R_f 0.30 (MeOH- EtOAc, 5:95); V_{max} (film/cm⁻¹) 3026, 2924, 1855, 1630, 1546, 1520, 1254, 799 and 780; $\delta_{\rm H}$ (400 MHz, CDCl $_{\rm 3}$) 9.09 (1H, s, H-6, triazine), 8.80-8.79 (1H, m, pyridyl), 8.52 (1H, d, J = 8.0 Hz, pyridyl), 7.81 (1H, ddd, J = 1.8, 8.0, 8.0 Hz, pyridyl), 7.36 (1H, dd, J = 4.9, 8.0 Hz, pyridyl), 2.83 (1H, tt, J = 3.1, 11.9 Hz, cyclohexyl), 1.96 (2H, br d, J = 12.5 Hz, cyclohexyl), 1.82 (2H, td, J = 3.1, 13.2 Hz, cyclohexyl), 1.69 (1H, br d, J = 12.5 Hz, cyclohexyl), 1.54 (4H, dq, J = 3.1, 12.5 Hz, cyclohexyl), 1.39-1.12 (3H, m, cyclohexyl); $\delta_{\rm C}$ (100 MHz, CDCl₃) 166.4 (q), 162.1.9 (q), 152.7 (q), 150.1 (CH), 147.3 (CH), 136.7 (CH), 125.1 (CH), 123.8 (CH), 43.9 (CH), 31.1 (2 x CH₂), 25.5 (2 x CH₂), 25.2 (CH₂); m/z (CI) 241 ([M+H]⁺, 100%) [HRMS (CI): calcd. for $C_{14}H_{16}N_4$, 241.1453. Found: $[M+H]^+$, 241.1453 (0.0 ppm error)].

2.2.18 Synthesis of 3-(2-pyridyl)-5-(2-furyl)-1,2,4-triazine 209c

To a 10 mL CEM Discover reaction vial with a stir bar was placed amidrazone **205** (0.040 g, 0.3 mmol), 1-(2-furyl)-2-hydroxyketone **206f** (0.037 mL, 0.3 mmol), MnO₂ (0.026 g, 0.3 mmol) and DCM (0.5 mL). The reaction vessel was irradiated for 60 minutes at 55 °C (power 100 W, pressure up to 300 psi). After 60 minutes, the vessel was cooled to 50 °C and the residue filtered through Celite ®, washed with DCM, concentrated in vacuo and purified by flash column chromatography (MeOH-EtOAc, 5:95) to give the title compound 209c as a dark yellow solid (0.037 g, 56%), mp 93.0-94.0 °C; R_f 0.35 (MeOH-EtOAc, 5:95); V_{max} (film/cm $^{-1}$) 1588, 1534, 1440, 1405, 1136, 1048, 1015 and 771; $\delta_{\rm H}$ (400 MHz, CDCl $_{
m 3}$) 9.47 (1H, s, H-6, triazine), 8.80 (1H, d, J = 4.0 Hz, pyridyl), 8.53 (1H, d, J = 7.9 Hz, pyridyl), 7.83 (1H, ddd, J = 1.5, 7.9, 7.9 Hz, pyridyl), 7.66 (1H, brs, furyl), 7.59 (1H, d, J = 3.4 Hz, furyl), 7.39 (1H, dd, J = 4, 7.9 Hz, pyridyl), 6.60-6.59 (1H, m, furyl); δ_{c} (100 MHz, CDCl₃) 162.4 (q), 152.7 (q), 150.3 (CH), 148.8 (q), 147.3, (q), 147.1 (CH), 143.3 (CH), 137.0 (CH), 125.5 (CH), 124.0 (CH), 116.9 (CH), 113.3 (CH); m/z (CI) 225 ([M+H] $^{+}$, 100%) [HRMS (CI): calcd. for $C_{12}H_0N_4O$, 225.0776. Found: [M+H] $^{+}$, 225.0771.

2.2.19 Synthesis of 3-(2-pyridyl)-5-cyclohexyl-1,2,4-triazine 209d

To a 10 mL CEM Discover reaction vial with a stir bar was placed amidrazone **205** (0.040 g, 0.3 mmol), hydroxyacetophenone **206g** (0.040 mL, 0.3

mmol), MnO₂ (0.026 g, 0.3 mmol) and DCM (0.5 mL). The reaction vessel was irradiated for 95 minutes at 55 °C (power 100 W, pressure up to 300 psi). After 95 minutes, the vessel was cooled to 50 °C and the residue filtered through Celite , washed with DCM, concentrated *in vacuo* and purified by flash column chromatography (MeOH-EtOAc, 5:95) to give the title compound **209d** as a yellow solid (0.027 g, 40%), mp 136.3-137.0 °C; R_f 0.40 (MeOH-EtOAc, 1:9); $\delta_{\rm H}$ (400 MHz, CDCl₃) 9.68 (1H, s, H-6, triazine), 8.89 (1H, d, d) = 4 Hz, pyridyl), 8.64 (1H, d), d) = 7.9 Hz, pyridyl), 8.26 (1H, d), d) = 8.0 Hz, aromatic), 7.92 (1H, ddd), d) = 1.8, 7.9, 7.9 Hz, pyridyl), 7.57-7.45 (3H, d), d) aromatic), 7.46 (1H, d), d) = 4.0, 7.9 Hz, pyridyl); $\delta_{\rm C}$ (100 MHz, CDCl₃) 162.6 (q), 155.6 (q), 152.8 (q), 150.3 (CH), 145.2 (CH), 137.0 (CH), 133.2 (q), 132.5 (CH), 129.3 (2 x CH), 127.7 (2 x CH) 125.5 (CH), 124.0 (CH).

2.2.20 Synthesis of aldehyde amidrazone 210a

To a solution of 2-pyridylamidrazone **205** (0.137 g, 1.0 mmol) in EtOH (5 mL) was added hydroxy acetone **206a** (0.1 mL, 1.4 mmol) and the mixture was stirred at room temperature for 2 hours. The reaction mixture was concentrated *in vacuo*. The second step, the crude condensation product was added powdered 4 Å molecular sieves (0.300 g), MnO₂ (0.106 g, 1.2 mmol) in DCM (6 mL) and the mixture heated at reflux for 1 hour and monitored by TLC. which showed that the starting material remained so a second portion of MnO₂ (0.273 g, 3.1 mmol) was added and continue heating at reflux for 1 hour followed by addition of a third portion of MnO₂ (0.454 g,

5.0 mmol) with heating at reflux for 2 hours. The reaction mixture was refluxed for a further 16 hours, cooled, filtered through Celite® and washed well with DCM. The combined organics were concentrated *in vacuo*. The crude product was purified by flash column chromatography (DCM) to give the title compound **210a** as a yellow solid (0.137 g, 79%), mp 147.8-148.9 °C; R_f 0.40 (PE-EtOAc, 1:1); V_{max} (film/cm⁻¹) 3416, 3280, 1677, 1513, 1264, 171, 1029, 855, 800 and 737; δ_{H} (400 MHz, CDCl₃) 9.61 (1H, s, aldehyde), 8.62 (1H, d, d = 4.0 Hz, pyridyl), 8.45 (1H, d, d = 7.9 Hz, pyridyl), 7.82 (1H, ddd, d = 1.8, 7.9, 7.9 Hz, pyridyl), 7.44 (1H, dd, d = 4.0, 7.9 Hz, pyridyl), 2.18 (3H, s, pyridyl); δ_{C} (100 MHz, CDCl₃) 194.6 (CH), 161.3 (q), 158.4 (q), 149.8 (q), 148.9 (CH), 137.1 (CH), 126.2 (CH), 122.6 (CH), 10.6 (CH₃), m/z: (CI) 191 ([M+H]⁺, 100 %) [HRMS (CI): calcd. for $C_9H_{11}N_4O$, 191.0932. Found: [M+H]⁺, 191.0934.

2.2.21 Synthesis of aldehyde amidrazone 210c

$$\operatorname{HO}_{N} \operatorname{H}_{2} \operatorname{N}_{N}$$

To a solution of 2-pyridylamidrazone **205** (0.155 g, 1.1 mmol) in EtOH (5 mL) was added 1,3-dihydroxyketone (dimer) **206e** (0.112 g, 0.6 mmol) and the mixture was stirred at room temperature for 1 hour and the reaction mixture concentrated *in vacuo*. The second step, to the crude condensation product was added powdered 4 Å molecular sieves (0.307 g), MnO₂ (0.166 g, 1.9 mmol) and DCM (10 mL) and the mixture was heated to reflux for 1 hour and monitored by TLC. which showed that the starting material remained so a second portion of MnO₂ (0.179 g, 2.0

mmol) was added, heated at reflux for 1 hour followed by addition of a third portion of MnO₂ (0.319 g, 3.6 mmol) with heating at reflux for 3 hours. A fourth portion of MnO₂ (0.200 g, 2.3 mmol) was added and the reaction mixture was refluxed for a further 24 hours, cooled, filtered through Celite[®] and washed well with DCM. The combined organics were concentrated *in vacuo*. The crude product was purified by flash column chromatography (MeOH-EtOAc, 5:95) to give the title compound **210c** (0.102 g, 52%), mp 125.4-127.1 °C; Rf 0.34 (PE-EtOAc, 1:1); V_{max} (film/cm⁻¹) 3405, 3283, 1677, 1640, 1513, 1407, 1275, 1177, 1083, 802 and 738; δ_{H} (400 MHz, CDCl₃) 9.51 (1H, s, aldehyde), 8.6 (1H, d, d = 4.0 Hz, pyridyl), 8.18 (1H, d, d = 7.9 Hz, pyridyl), 7.79 (1H, ddd, d = 1.8, 7.9, 7.9 Hz, pyridyl), 7.42 (1H, dd, d = 4.0, 7.9 Hz, pyridyl), 4.71 (3H, s); δ_{C} (100 MHz, CDCl₃) 192.3 (CH), 159.9 (q), 159.4 (q), 148.9 (CH), 148.1 (q), 137.14 (CH), 126.6 (CH), 122.3 (CH), 57.9 (CH₂, C-7'); m/z (CI) 207 ([M+H]⁺, 100 %) [HRMS (CI): calcd for: $C_9H_{11}N_4$ O_2 , 207.0882. Found: [M+H]⁺ 207.0877.

2.2.22 Synthesis of 3-methyl-1-(2-pyrydyl)-6,7-drihydro-5H-[2]pyrindine 211b

Triazine **209a** (0.017 g, 0.1 mmol), cyclopentanone (10 µL, 0.11 mmol), *N*-methylethylenediamine (10 µL, 0.12 mmol) and toluene (0.1 mL) were placed in a 10 mL CEM Discover reaction vial and irradiated at 150 °C for 60 minutes (power 300 W, pressure up to 300 psi). After 60 minutes, the vessel was cooled down to 50 °C and diluted with DCM followed by concentration *in vacuo* and purification by flash

column chromatography (EtOAc) to give the title compound **211b** as a yellow viscous oil (0.008 g, 41%), R_f 0.33 (EtOAc); \mathbf{V}_{max} (film/cm⁻¹) 2955, 1587, 1564, 1471, 1418 and 744; δ_{H} (400 MHz, CDCl₃) 8.61 (1H, d, J = 4.9 Hz, pyridyl), 8.07 (1H, d, J = 7.9, pyridyl), 7.72 (1H, dd, J = 7.9, 7.9 Hz, pyridyl), 7.19 (1H, dd, J = 4.9, 7.9 Hz, pyridyl), 7.03 (1H, s, H-3), 3.22 (2H, t, J = 7.6 Hz), 2.85 (2H, t, J = 7.6), 2.55 (3H, s), 2.01 (2H, guin, J = 7.3); m/z (CI): 211 ([M+H]⁺, 100%) [HRMS (CI): calcd. for $\mathbf{C}_{14}\mathbf{H}_{15}\mathbf{N}_{2}$, 221.1235. Found: [M+H]⁺, 221.1237.

2.2.23 Synthesis of 3-cyclohexyl-1-(2-pyrydyl)- 6,7-drihydro-5H-[2]pyrindine 211d

Triazine **209b** (0.025 g, 0.1 mmol), cyclopentanone (10 µL, 0.11 mmol), *N*-methylethylenediamine (10 µL, 0.12 mmol) and toluene (0.1 mL) were placed in a 10 mL CEM Discover reaction vial and irradiated at 150 °C for 90 minutes (power 300 W, pressure up to 300 psi). After 90 minutes, the vessel was cooled down to 50 °C and diluted with DCM followed by concentration *in vacuo* and purification by flash column chromatography (PE-EtOAc, 1:1) to give the title compound **211d** as a yellow viscous oil (0.010 g, 33%); R_f 0.45 (PE-EtOAc, 4:1); \mathbf{V}_{max} (film/cm⁻¹) 2925, 2851, 1586, 1563, 1470, 1449, 1417, 1255, 1096, 802 and 744; δ_{H} (400 MHz, CDCl₃) 8.58 (1H, d, d = 4.0 Hz, pyridyl), 8.16 (1H, d, d = 7.9 Hz, pyridyl), 7.71 (1H, dd, d = 7.9, 7.9 Hz, pyridyl), 7.19 (1H, s, H-3), 7.16 (1H, dd, d = 4.0, 7.9 Hz, pyridyl), 3.25 (2H, t, 7.6 Hz), 2.84 (2H, t, d = 7.6 Hz), 2.71 (1H, d = 3.1, 8.6 Hz), 2.00 (2H, d = 7.6 Hz), 1.95-1.18 (10H, d); δ_{C} (100 MHz, CDCl₃) 163.7 (q), 158.7 (q), 156.1 (q),150.7

(q), 148.4 (CH), 136. (q), 136.3 (CH), 123.1 (CH), 122.4 (CH), 116.9 (CH), 46.2 (CH) 33.2 (2 x CH₂), 32.6 (CH₂), 32.5 (CH₂), 26.6 (2 x CH₂), 26.1 (CH₂), 25.0 (CH₂); m/z (CI): 279 ([M+H]⁺, 100%) [HRMS (CI): calcd. for $C_{19}H_{23}N_2$, 279.1861. Found: [M+H]⁺, 279.1862.

2.2.24 Synthesis of 4-furyl-1-(2-pyridyl)- 6,7-drihydro-5H-[2]pyrindine 211h

Triazine **208f** (0.012 g, 0.05 mmol), cyclopentanone (4 µL, 0.05 mmol), *N*-methylethylenediamine (4 µL, 0.05 mmol) and toluene (0.05 mL) were placed in a 10 mL CEM Discover reaction vial and irradiated at 170 °C for 15 minutes (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled down to 50 °C. It was then diluted with DCM, followed by concentration *in vacuo* and purification by flash column chromatography (EtOAc) to give the title compound **211h** as a yellow solid (0.010 g, 71%), mp 96.5-97.8 °C; R_f 0.20 (EtOAc); V_{max} (film/cm⁻¹) 1638, 1429, 1091 and 744; δ_{H} (400 MHz, CDCl₃) 8.84 (1H, s, H-3), 8.62 (1H, d, d = 3.4 Hz, pyridyl), 8.15 (1H, d, d = 8.0 Hz, pyridyl), 7.75 (1H, d, d = 8.0 Hz, pyridyl), 7.51 (1H, d), d = 3.4 Hz, furyl), 6.48 (1H, d), d = 7.6 Hz), 3.11 (2H, d), d = 7.6 Hz), 2.10 (2H, d), d = 7.6 Hz); d (CI): 263 ([M+H] $^+$, 100%) [HRMS (CI): calcd. for $C_{17}H_{15}N_2O$, 263.1184. Found: [M+H] $^+$, 263.1187.

2.2.25 Synthesis of 3-furyl-1-(2-pyridyl)- 6,7-drihydro-5*H*-[2]pyrindine 211i

Triazine 209d (0.022 g, 0.1 mmol), cyclopentanone (8 µL, 0.1 mmol), Nmethylethylenediamine (8 µL, 0.1 mmol) and toluene (50 µL) were placed in a 10 mL CEM Discover® reaction vial and irradiated at 170 °C for 15 minutes (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled down to 50 °C, diluted with DCM and then concentrated in vacuo and purified by flash column chromatography (PE-EtOAc, 1:1) to give the title compound 211i as a yellow solid (0.015 g, 60%), mp79.5-81.6 °C; R_f 0.42 (PE-EtOAc, 1:4); V_{max} (film/cm⁻¹) 2954, 1605, 1585, 1564, 1494, 1472, 1417, 1391, 1006 and 738; $\delta_{\!\scriptscriptstyle H}$ (400 MHz, CDCl $_{\!\scriptscriptstyle 3}$) 8.60 (1H, d, J = 4.0 Hz, pyridyl), 8.26 (1H, d, J = 7.9 Hz, pyridyl), 7.74 (1H, ddd, J = 1.8, 7.9, 7.9 Hz, pyridyl), 7.55 (1H, s, H-4), 7.45 (1H, d, J = 0.9 Hz, furyl), 7.22-7.20 (1H, m, pyridyl), 7.01 (1H, d, J = 3.4 Hz, furyl), 6.45 (1H, dd, J = 0.9, 3.4 Hz, furyl), 3.33 (2H, t, J = 7.6 Hz), 2.91 (2H, t, J = 7.6 Hz), 2.04 (2H, quin, J = 7.6 Hz); $\delta_{\rm C}$ (100 MHz, CDCl₃) 158.2 (q), 156.5 (q), 154.3 (q), 151.4 (q), 148.3 (CH), 146.8 (q), 142.6 (CH), 137.7 (q), 136.4 (CH), 123.1 (CH), 122.7 (CH), 114.7 (CH), 111.8 (CH), 107.7 (CH), 32.9 (CH₂), 32.7 (CH₂), 24.9 (CH₂); m/z (CI): 263 ([M+H]⁺, 100%) [HRMS (CI): calcd. for $C_{17}H_{15}N_2O$, 263.1184. Found: $[M+H]^+$, 263.1187...

One pot reaction: 1-(2-Furyl)-2-hydroxy ketone **206f** (0.037 g, 0.3 mmol), 2-pyridylamidrazone **205** (0.040 g, 0.3 mmol), cyclopentanone (26 μ L, 0.3 mmol), *N*-methylethylenediamine (26 μ L, 0.3 mmol), MnO₂ (0.026 g, 0.3 mmol) and toluene (50

µL) were placed in a 10 mL CEM Discover reaction vial and irradiated at 170 °C for 15 minutes (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled down to 50 °C, diluted with DCM and then concentrated *in vacuo* and purified by flash column chromatography (PE-EtOAc, 1:1) to give the title compound **211i** as a yellow solid (0.011 g, 15%).

2.2.26 Synthesis of 4-phenyl-1-(2-pyrydyl)- 6,7-drihydro-5*H*-[2]pyrindine 211j

Triazine **208g** (0.023 g, 0.1 mmol), cyclopentanone (8 µL, 0.1 mmol), *N*-methylethylenediamine (8 µL, 0.1 mmol) and toluene (0.1 mL) were placed in a 10 mL CEM Discover reaction vial and irradiated at 170 °C for 15 minutes (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled down to 50 °C, diluted with DCM and then concentrated *in vacuo* and purified by flash column chromatography (EtOAc) to give the title compound **211j** (0.015 g, 55%) as a yellow solid, mp 121.3-122.8 °C; R_f 0.25 (PE- EtOAc, 4:1); V_{max} (film/cm⁻¹) 3056, 2956, 1583, 1552, 1445, 1428, 1375, 1088, 766, 745, 723 and 700; δ_{H} (400 MHz, CDCl₃) 8.63 (1H, d, d = 4.3 Hz, pyridyl), 8.48 (1H, s, H-3), 8.12 (1H, d, d = 8.2 Hz, pyridyl), 7.75 (1H, ddd, d = 1.2, 8.2, 8.2 Hz, pyridyl), 7.45-7.31 (5H, m, aromatic), 7.22 (1H, ddd, d = 1.2, 4.28, 8.2 Hz, pyridyl), 3.38 (2H, d = 7.3 Hz), 2.97 (2H, d = 7.3 Hz), 2.01 (2H, d = 7.3 Hz); δ_{C} (100 MHz, CDCl₃) 157.9 (q), 153.3 (q), 150.6 (q), 148.6 (2 x CH), 146.4 (CH), 139.3 (q), 137.6 (q), 136.3 (2 x CH), 133.6 (q), 128.5 (2 x

CH), 127.6 (CH), 122.9 (CH), 122.6 (CH), 33.3 (CH₂), 32.5 (CH₂), 25.4 (CH₂); m/z (CI): 273 ([M+H]⁺, 100%) [HRMS (CI): calcd. for $C_{19}H_{17}N_2$, 273.1391. Found: [M+H]⁺, 273.1393.

2.2.27 Synthesis of 3-phenyl-1-(2-pyrydyl)- 6,7-drihydro-5H-[2]pyrindine 211k

Triazine **209e** (0.023 g, 0.1 mmol), cyclopentanone (8 µL, 0.1 mmol), *N*-methylethylenediamine (8 µL, 0.1 mmol) and toluene (0.1 mL) were placed in a 10 mL CEM Discover reaction vial and irradiated at 150 °C for 60 minutes (power 300 W, pressure up to 300 psi). After 15 minutes, the vessel was cooled down to 50 °C, diluted with DCM and then concentrated *in vacuo* and purified by flash column chromatography (PE-EtOAc, 4:1) to give the title compound **211k** (0.018 g, 67%) as a white solid, mp 106.3-107.0 °C; R_f 0.37 (PE- EtOAc, 4:1); $\delta_{\rm H}$ (400 MHz, CDCl₃) 8.62 (1H, d, J = 4.0 Hz, pyridyl), 8.37 (1H, d, J = 7.9 Hz, pyridyl), 8.04 (2H, d, J = 7.0 Hz, aromatic), 7.76 (1H, ddd, J = 1.5, 7.9, 7.9 Hz, pyridyl), 7.60 (1H, s, H-4), 7.40 (2H, t, J = 7.0 Hz, aromatic and pyridyl), 7.32 (2H, t, J = 7.0 Hz, aromatic), 3.38 (2H, t, J = 6.7 Hz), 2.94 (2H, t, J = 6.7 Hz), 2.07 (2H, t, t), 139.8 (q), 137.9 (q), 136.4 (CH), 128.5 (q), 156.8 (q), 154.4 (q), 151.1 (q), 148.3 (CH), 139.8 (q), 137.9 (q), 136.4 (CH), 128.5 (2 x CH), 128.4 (CH), 126.8 (2 x CH), 123.1 (CH), 122.6 (CH), 116.5 (CH), 32.9 (CH₂), 32.7 (CH₂), 25.0 (CH₃).

2.3 RESULTS AND DISCUSSION

The tethered imine-enamine methodology has been developed for the direct conversion of 1,2,4-triazines into highly substituted pyridine 211 *via* the inverse electron demand Diels-Alder reaction. The synthetic route to highly substituted pyridines 211 was outline in **Scheme** 43. This methodology employed the condensation reaction of precursor amidrazone 205 and alcohol ketones 206 as a key step, to give the corresponding 1,2,4-triazines 208 and 209 which were directly converted into highly substituted pyridines 211.

Scheme 43 Microwave assisted synthesis of highly substituted pyridines 211

R = alkyl, furyl or phenyl

2.3.1 Microwave assisted regioselective synthesis of 6-substituted 1,2,4-triazines 208a-e

6-Substituted 1,2,4-triazines 208a-e could be obtained via the amidrazone 207a-e which were prepared from amidrazone 205 (Case, 1965), and hydroxyl alcohols 207a-e (Moriarty et al., 1992). Compounds 207a-e were obtained from direct reaction between amidrazone 205 and alcohols 206a-e in EtOH at room temperature for 2-4 hours, to give the corresponding compounds 207a-e in excellent yield (80-93%, Table 28, Scheme 44). The next step was oxidation of the 1°-alcohol of compound 207a-e to the aldehyde (without isolation) which could be achieved by the reaction of 207a-e with MnO₂ in toluene and irradiated at 120 °C for 10-30 minutes. After cooling, filtering and concentration, the crude product was further irradiated at 150 °C for 10 minutes in toluene and in the presence of AcOH to give the 1,2,4-triazines 208a-e in moderate to excellent yield (55-98%, Table 28, Scheme 44)

a: R = methyl; b: R = cyclohexyl; c: R= pentyl; e: R = methylene hydroxy

a) MnO₂, PhCH₃, MW, 120 °C, 5-30 mins, b) PhCH₃, AcOH, MW, 150 °C, 10 mins, c)DCM 2 mL/ AcOH 10 µL for **207a**, 48 hrs CHCl₃ 2mL for **5c**, 24 hrs

Scheme 44 Microwave assisted synthesis of 6-substituted 1,2,4-triazines 208a-e (route A) and conventional heating synthesis of 208a and 208c (route B)

Entry	Compound	5 (% yield)	6 (% yield)
1	он 206а	207a (93)	208a (98) ^a
2	он 206b	207b (88)	208b (97) ^a
	206c +	207c +	208c (55) ^b
3	206d	207d (80, mixtures)	208d (63) ^b
4	но он 206е	HO H ₂ N N	HO N N
		207e (80)	208e (95) ^c

Table 28 Microwave assisted synthesis of 6-substituted 1,2,4-triazines 208a-e

The reaction of **207a** and **207e** *via* conventional heating needed reaction time more than those of microwave heating and the yields were slightly less than the reaction *via* microwave heating.

The condensation reaction of hydroxyl alcohols **206f** and **206g** with amidrazone **205** were not successful *via* conventional heating. This may be due to the

^aThe yield without isolation. ^bIsolated yield and calculation from the ratio of the alcohol **206c** and **206d** (1.6:1).

^cThe product was derived from the aldehyde **210b** and irradiated at 150 °C in toluene (less trizine **208e** was observed when compound **207e** used as starting material)

less reactivity of carbonyl functionality which were stabilized by the aromatic ring system of both of phenyl and 2-furyl moieties and probably this reaction needed higher temperature than those of hydroxyl alcohols **206a-e**. Fortunately, these reactions were successful *via* microwave heating. The irradiation at 90 °C for 10-40 minutes of hydroxyl alcohols **206f** or **206g** and amidrazone **205** in EtOH gave the corresponding intermediates **207f** and **207g** (detected from TLC checked and without isolation). After cooling down to 50 °C and concentration, the crude products were immediately oxidized with MnO₂ in toluene and irradiated at 120 °C for 15-40 minutes to yield the 1,2,4-triazines **208f** (22 %) and **208g** (50%) (**Scheme** 45).

Scheme 45 Microwave assisted synthesis of 6-substituted triazine derivatives 208f and 208g

2.3.2 Microwave assisted regioselective synthesis of 5-substituted 1,2,4-triazines 209a-e

The synthesis of regioselective of 5-substituted 1,2,4-triazine would be derived from one pot reaction of amidrazone, hydroxyl ketones and MnO₂. The reaction mixture of amidrazone, hydroxyl ketones and MnO₂ in DCM was irradiated at 55 °C for 55-95 minutes to give the triazine **209a-d**. As can be seen that the one pot reaction was obtained in moderate yield for these cascade reactions; the long sequence being condensation, oxidation and cyclization to the triazines **209a-d** in overall yields of 45%, 38%, 56% and 40%, respectively (**Scheme** 46).

Scheme 46 Microwave assisted synthesis of 5-substituted triazine derivatives 209a-d

2.3.3 Microwave assisted synthesis of 3 and 4-substituted of 6,7-Dihydro-5*H*-[2]pyrindines 211a-h

The conversion of 1,2,4-triazine into [2]pyrindine compounds used the Boger's methodology. This chemistry exploits the inverse electron demand Diels-Alder reaction of triazines and enamine. [2]Pyrindins 211a-h would be obtained from triazines 208a-g and 209a-d. The reactions of triazines 208a-g or 209a-d and cyclopentanone in toluene in the presence of *N*,*N*-methylethyldiamine were irradiated at 145-170 °C for 30-60 minutes which led to the formation of [2]pyrindins 211a-h in moderate to good yields (Scheme 47, Table 29). The mechanism of this reaction was shown in Scheme 48. In the first step, the cyclopentanone was reacted with *N*,*N*-methylethyldiamine to give the intermediate emine-enamine (212) which was immediately treated with 1,2,4-triazine yielding the intermediate (213). The intramolecular cyclization of the intermediate afforded dihydropyridine 211 which could exist as a zwitterion 215. Finally, the zwitterion would undergo elimination *in situ*, leading directly to [2]pyrindines.

Scheme 47 Microwave assisted synthesis of 3 and 4-substituted of 6,7-Dihydro-5*H*-[2]pyrindines **211a-h**

Scheme 48 The mechanism of the formation of 3- and 4-substituted of 6,7-Dihydro-5*H*-[2]pyrindines **211a-h**

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The substitutents at C-5 or C-6 of 1,2,4-triazines showed an effect in the inverse electron demand Diels-Alder reaction. For example, if the substituted groups were phenyl or 2-furyl moieties (both position C-5 and C-6), the reaction gave the best results, moderate to good yield with the yield 71%, 60%, 65% and 67% for [2]pyrindin 211h, 211i, 211j and 211k, respectively. While the alkyl groups at C-5 or C-6 of 1,2,4-triazines exhibited less reactive than those of phenyl or 2-furyl groups at the same position (Table 29). It is interesting to note that the C-5 alkyl substituted 1,2,4-triazines with 41% and 33% isolated yield for 211b and 211c, respectively. No reaction was observed when using C-6 alkyl substituted 1,2,4-triazines as starting material (Table 29). Finally, the one pot reaction of 1-(2-furyl)-2-hydroxy ketone, amidrazone, cyclopentanone, *N,N*-methylethyldiamine, MnO₂ in toluene was irradiated at 170 °C

for 15 minutes to give the corresponding [2]pyrindine **211g**. The reaction was obtained in low yield for these cascade reactions, the long sequence being oxidation, double condensations, Diels-Alder, retro-Diels-Alder and aromatization to **211g** in overall yield of 15% from **206f** (**Scheme** 49). The ¹H NMR spectra of **208**, **209** and **211** were summarized in appendix.

Table 29 Microwave assisted synthesis of 3- and 4-substituted of pyridines 211a-h

Entry	Triazine	211 yield (%)	Entry	Triazine	211 yield (%)
1	208a	a (0)	7	но	BO N N N N N N N N N N N N N N N N N N N
2	209a	b (41)	8	208f	h (71)
3	208b	c (0)	9	209c	i (60)
4	209b	d (33)	10	208g	j (55)
5	208c	e (less)	11	209d	k (67)
6	208d	f(0)			

Scheme 49 Microwave assisted synthesis of 3-(2-furyl)-1-(2-pyridyl)-6,7-dihydro-5*H*-[2]pyrindine **211i**

In conclusion, the direct conversion of 1,2,4-triazines **208** or **209** into highly substituted pyridines **211** was developed which eliminated the need for a second, discrete aromatization step. The methodology was operationally simple and afforded pyridines **211** in moderate to good yields.